

7 α ,11 β -DIACETOXYDIHYDRONOMILIN,
A NEW TETRANORTRITERPENOID FROM CEDRELA MEXICANA

G.B. Marcelle¹ and B.S. Mootoo^{*}
Department of Chemistry, University of the West Indies,
St. Augustine, Trinidad, W.I.

Abstract: Spectroscopic and chemical evidence is presented in support of the structure of 7 α ,11 β -diacetoxydihydronomilin (I) obtained from a benzene extract of the dehisced fruit capsules of Cedrela mexicana.

A variety of related tetranortriterpenoids has been reported from the heartwood of Cedrela mexicana collected in Jamaica², Trinidad³ and West Africa⁴.

We wish to report a new compound 7 α ,11 β -diacetoxydihydronomilin (I) from the dehisced fruit capsules of this plant collected in Trinidad. This is a unique addition to the increasing number of seco-ring A tetranortriterpenoids recently reported, e.g. 7 α -acetoxydihydronomilin⁵, tricoccin S₁₃⁶, evodulone⁷ and surenone and surenin⁸.

Isolation:

The benzene extract of 2.3 kg of ground material yielded 31.2 g of a gummy residue from which was obtained 10.5 g of a petroleum ether insoluble precipitate. Repeated PLC on silica gel with acetone/petroleum ether (1:2) yielded a crystalline compound m.p. 240° - 45° (ethyl acetate/petroleum ether) which analysed for C₃₂H₄₀O₁₂, [calc. C 62.32, H 6.54, found C 62.07, H 6.55]. Crystallisation from methanol gave m.p. 262 - 64° with methanol of crystallisation.

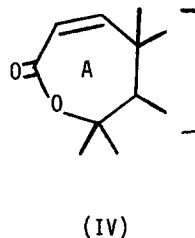
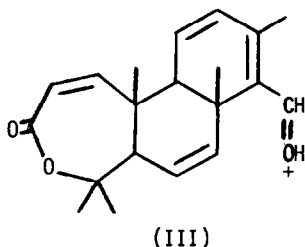
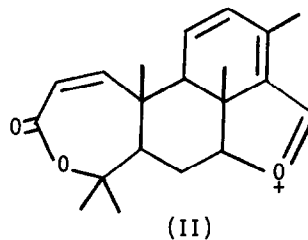
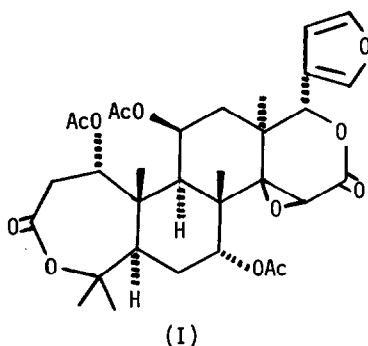
Structure:

The IR (nujol) spectrum showed a broad band centred at 1740 cm⁻¹ (ester, lactone) and one at 870 cm⁻¹ (β -substituted furan). The ¹H-NMR (270MHz, CDCl₃, δ ppm) confirmed the β -substituted furan (6.30, 7.39, 7.43, 1H each) and further indicated five tertiary methyl groups (1.20, 1.35, 1.39, 1.44, 1.50) and three secondary acetate groups (2.06, 2.10, 2.12).

The corresponding protons on the acetate bearing carbons could be located as follows - C₁-H, 4.85 (d, J=6.5Hz), C₇-H, 4.53(s)⁹ and the C₁₁-H, 5.19 (quintuplet, $w^{\frac{1}{2}} = 21\text{Hz}$)¹⁰. The assignment of the C₁-proton was further confirmed by double irradiation at 3.20 when it collapsed to a singlet. The C₁₅-H was located at 5.35(s) and the C₁₇-H at 5.60(s).

In addition the remaining eight protons in the molecule could be assigned as follows: the C₂-protons, 3.20 (2H, m); C₉-H, 2.89 (1H, d); C₅-H, 2.45 (1H, quartet); C₁₂-H, 2.32 (1H, quartet); C₁₂-H, 1.92 (1H, quartet) and the two C₆ protons, 1.63 (2H, s).

The C₁₇-H at 5.60 together with the C₁₅-H at 3.60 confirmed the epoxy- δ -lactone structure for ring D in combination with a C₇- α -OAc¹¹.



The remaining two oxygen atoms are accounted for by the ring A lactone structure. The latter is supported by the following evidence:-

- (a) the low field positions of the five tertiary methyl groups (1.20 - 1.50);
- (b) the peak in the ms at m/e 255 which can be obtained from either of the readily formed ions of m/e 313, (II) or (III) by loss of acetone from ring A¹².

The structure of ring A was confirmed by the treatment of compound (I) with Al₂O₃ (neutral, super Grade I, Woelm). When the compound was stirred with Al₂O₃ at RT in benzene for 2.5 hours the product obtained was the α,β -unsaturated lactone (IV) - λ_{max} [EtOH] 215 nm

(ϵ , 16,000) compared with compound (I) - λ_{\max} [EtOH] 213nm (ϵ , 6,000); ν_{\max} (Nujol) - 1740 cm^{-1} , shoulder at 1690 cm^{-1} ; $^1\text{H-NMR}$ (60MHz, CDCl_3 , δppm) - two secondary acetates (2.01, 2.10) with the corresponding protons at 4.57 (broadened singlet) and 5.60 (m), the latter overlapping with the $\text{C}_{15}\text{-H}$. Two new doublets corresponding to the C_1 and C_2 protons of the α,β -unsaturated lactone were centred at 6.30 (d, $J=13\text{Hz}$) and 5.90 (d, $J=13\text{Hz}$) respectively.

The -OAc group at C_1 has been assigned an α -configuration. Hydroxy or acetyl functions at this position in limonoids are usually α ¹³ and an X-ray structure has recently confirmed this configuration in 7α -acetoxydihydronomilin⁵. Dreyer⁹ had proposed a β -configuration in nomilin for which compound he reported a quartet for the C_1 proton (60MHz). We also obtained a quartet using a 60MHz instrument for the $\text{C}_1\text{-H}$ in nomilin and what appeared to be a badly resolved triplet for the corresponding proton in compound (I). Ahmed et al⁵ have quoted the two protons on the acetate bearing carbons as multiplets at 4.84 and 4.50 in 7α -acetoxydihydronomilin.

When measured on a 270MHz instrument the $\text{C}_1\text{-H}$ of nomilin appeared as a distinct doublet at 5.04 ($J=6.5\text{Hz}$), coincided in shape with the $\text{C}_1\text{-H}$ of compound (I) (d, 4.85, $J=6.5\text{Hz}$), and like this proton in compound (I), collapsed to a singlet on irradiation at 3.20.

The seven-membered lactone ring A is very flexible and the shape of the C_1 proton in the NMR does not allow a conclusive assignment of the configuration at this carbon. However, if we assume the chair configuration of ring A in which $\text{C}_2\text{-C}_3\text{-O-C}_4$ are in the same plane (cf. X-ray evidence for 7α -acetoxydihydronomilin)⁵, an α -configuration for the $\text{C}_1\text{-OAc}$ will result in dihedral angles of approximately 30° and 90° between the $\text{C}_1\text{-}\beta$ -proton and the two protons at C_2 and account for the $\text{C}_1\text{-}\beta$ -proton in both nomilin and compound (I) appearing as a doublet ($J=6.5\text{Hz}$, 270MHz).

Investigations are continuing on the crude extract of the dehisced fruit capsules of Cedrela mexicana, since, in addition to $7\alpha,11\beta$ -diacetoxydihydronomilin, TLC shows other compounds which have not been previously isolated from this plant.

ACKNOWLEDGEMENT

We wish to thank Dr. Percy Manchand, Hoffman La Roche Inc. for MS; Professor J. Kutney, University of British Columbia for 270MHz $^1\text{H-NMR}$ and Messrs. C. Baldeo and S. Diljohn for technical assistance. One of us (GBM) acknowledges the award of the "Sir Frank Worrell Memorial Scholarship" from the University of the West Indies.

REFERENCES

1. Part of the M.Sc. Thesis of G. Marcelle - U.W.I. (1977).
2. W.R. Chan, K.E. Magnus and B.S. Mootoo, *J. Chem. Soc. (C)*, 171, (1967).
3. J.D. Connolly, R. McCrindle and K.H. Overton, *Tetrahedron*, 24, 1489, (1968).
4. C.W.L. Bevan, J.W. Powell and D.A.H. Taylor, *J. Chem. Soc.* 980, (1963).
5. F.R. Ahmed, Ang S. Ng and A.G. Fallis, *Can. J. Chem.* 56, 1020, (1978).
6. B. Epe and A. Mondon, *Tetrahedron Lett.*, 3901, (1978).
7. B.L. Sondengam, C.S. Kamga and J.D. Connolly, *ibid.*, 1357, (1979).
8. W. Kraus and K. Kypke, *ibid.*, 2715, (1979).
9. D.L. Dreyer, *Tetrahedron*, 21, 75, (1965).
10. J.D. Connolly, R. McCrindle, K.H. Overton and J. Feeney, *Tetrahedron*, 22, 891, (1966).
11. J.W. Powell, *J. Chem. Soc. (C)*, 1794, (1966).
12. D.R. Taylor, *Rev. Latino Amer. Quim.*, 2, 87, (1971).
13. J.D. Connolly and K.H. Overton, *Progress in Phytochemistry*, II, 383, (1970).

(Received in USA 26 July 1980)